

Modeling of the Influence of Transverse Cracking on Chloride Penetration into Concrete

Dale P. Bentz
Edward J. Garboczi
Yang Lu
Nicos Martys
Aaron R. Sakulich
W. Jason Weiss

Materials and Structural Systems Division
Engineering Laboratory
National Institute of Standards and Technology
100 Bureau Drive, Stop 8615
Gaithersburg, MD 20899-8615 USA
E-mail: dale.bentz@nist.gov
Phone: (301)975-5865

Abstract

Concrete service life models have proliferated in recent years due to increased interest in designing infrastructure elements with at least a 75-year service life, along with greater emphasis on life cycle costing in general. While current models consider a variety of concrete material and environmental factors, at varying levels of complexity, in predicting the time until the onset of chloride-induced corrosion of the steel reinforcement, the influence of cracking is generally beyond their current scope. This paper presents a preliminary strategy for examining the influence of transverse cracking on chloride ion penetration into concrete that includes a graphical approach for adjusting the predicted service life provided by current models to reflect this influence. Comparison to experimental data indicates that the contributions of binding of the chloride ions by the cement paste play a significant role in slowing the ingress of chlorides and should be accounted for in any modeling efforts.

Keywords: Binding, chloride diffusion, corrosion, service life, transverse cracking.

Introduction

Service life prediction and durability-based design are slowly entering the concrete design mainstream. The present focus on sustainability requires that accurate service life predictions be available for a concrete structure in its intended environment to enable life cycle cost analysis and to optimize sustainable performance. Numerous numerical models are available, some commercial and some free of charge, for performing service life prediction, generally considering chloride-induced corrosion of the steel reinforcement as the major degradation scenario. The complexity of these models vary widely, with some based on a straightforward Fick's second law-type analysis of chloride ion diffusion into saturated concrete, while others consider multi-species diffusion/convection into partially saturated concrete undergoing wetting/drying cycles. However, one ubiquitous limitation of existing models is their

inability to consider the influence of cracking on concrete service life without extracting cracked cores. Cracks provide preferential pathways for the ingress of deleterious species such as chloride ions and surely reduce the service life when present. But, by how much do cracks directly reduce service life?

Cracks can occur due to many causes (e.g., plastic shrinkage, restrained shrinkage, thermal and mechanical loading, improper design, expansive degradation reactions) and their three-dimensional geometry can take many forms. Perhaps the simplest case of practical significance to laboratory studies and field performance is the presence of transverse cracking, with the cracks often occurring directly above the top layer of steel reinforcement [1-3]. Such cracks can dramatically increase the buildup of chlorides at the depth of the steel reinforcement. In fact, results of Lindquist et al. [3] indicate that for cracked regions of bridge decks, sufficient chloride levels to initiate corrosion of the steel reinforcement can be reached after as little as one year of field exposure.

Transverse cracking can be modeled in a simplified form as shown in Figure 1, by considering a single crack (rectangular or triangular in shape) located directly above the steel reinforcement, a worst case scenario, but one that routinely occurs in the field [1]. When chlorides are present on the concrete surface, due to the application of de-icing salts or the concrete structure being located in a coastal region, chlorides will penetrate into both the bulk concrete and into the crack, when water-filled. The chloride transport rate will typically be much faster into the crack, and from the crack, the chlorides will spread into the surrounding concrete, effectively short-circuiting the slower route through the full (cover) depth of the bulk concrete to the reinforcement. While some research groups have analyzed this problem by determining effective diffusion coefficients for the crack and for the bulk concrete in a parallel flow-type model [4], Garces Rodriguez and Hooton [5] have pointed out that transport of chloride ions in cracked concrete should be analyzed as a two-dimensional problem, with a simple Fickian one-dimensional analysis being insufficient. In this case, both the crack width and crack depth are significant parameters influencing the overall chloride penetration, and relations between the crack width and the effective diffusivity of the crack have been developed in several cases [6-13]. Clearly, in the case shown in Figure 1, the concentration of chloride ions directly below the crack at the reinforcement surface will be higher than that obtained just a few millimeters to the right or to the left of the rebar. Since it is the local (and not the average) concentration of chlorides that is critical in inducing corrosion, this local value is what must be assessed in modeling for predicting service life.

This paper explores a variety of two-dimensional models for estimating the local chloride concentration as a function of crack geometry and uses these results to predict the anticipated reduction in reinforced concrete service life due to the presence of such transverse cracks. Such an approach was previously considered in the work of Paulsson-Tralla and Silfwerbrand in 2002 [6]. While their work estimated the increase in concrete cover necessary to mitigate a reduced service life as a function of crack depth, here the focus will be simply on how much the service life of the concrete is reduced due to cracking, as a function of crack geometry, reinforcement (cover) depth, and bridge deck concrete thickness.

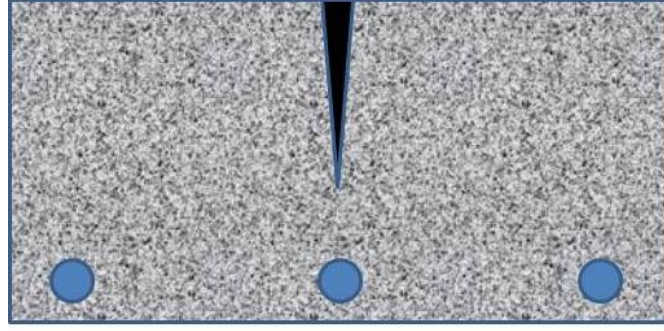


Figure 1. Schematic of transverse crack in concrete located directly above steel reinforcement (not to scale).

Theoretical Background and Computer Models

Solution to Fick's Second Law

For the case of one-dimensional diffusion with no binding/reaction, no chlorides initially present in the concrete, and a constant diffusion coefficient D , the following analytical solution for diffusion, according to Fick's 2nd law in a semi-infinite media, can be utilized to estimate the concentration of chloride ions as a function of depth and time [14]:

$$\frac{C(x,t)}{C_S} = \operatorname{erfc}\left(\frac{x}{2\sqrt{Dt}}\right) \quad (1)$$

where C_S is the chloride concentration at the top surface of the concrete, D is the chloride ion diffusion coefficient in the concrete, x is depth, t is time, and erfc indicates the complementary error function. In this form, the utility of plotting results as dimensionless variables, namely $C(x,t)/C_S$ vs. $x/(2\sqrt{Dt})$, is apparent. Equation 1 provides a convenient analytical solution to evaluate the validity of one-dimensional finite element or finite difference models for diffusive transport. Additionally, if the chloride concentration necessary to initiate corrosion at a known depth of reinforcement x is supplied, along with the concrete diffusion coefficient and the surface chloride concentration, equation 1 can be solved to determine the estimated service life, as the time t to achieve this concentration at depth x [15].

ANSYS¹

While typically used for heat transfer and mechanical stress problems, ANSYS has also recently been applied to consider diffusion in cracked concrete [9,10]. This is usually accomplished by establishing an analogy between mass transfer and heat transfer; free chloride content is mapped to temperature, while the diffusion coefficient for the chloride ions is mapped to thermal conductivity, with the corresponding heat capacity and density parameters set to 1 [10]. The diffusion coefficient (thermal conductivity) can be further modified to be a

¹ Certain commercial products are identified in this paper to specify the materials used and procedures employed. In no case does such identification imply endorsement or recommendation by the National Institute of Standards and Technology, nor does it indicate that the products are necessarily the best available for the purpose.

function of time and temperature [10] or converted to an effective diffusion coefficient that includes equilibrium binding of chloride ions by the hydrated cement paste in the concrete based on various isotherm equations (linear, Langmuir, or Freundlich, etc.) [9]. Such binding typically includes the combined effects of sorption of the Cl^- ions by the calcium silicate hydrate gel (C-S-H) and other hydration products, as well as the reaction of these ions with aluminate phases present in the hydrated cement paste to form Friedel's salt (and perhaps other chloroaluminate compounds). However, incorporating the kinetics of the binding process is a more challenging problem that has not yet been effectively addressed. It is well recognized that binding has a considerable influence on the rate of chloride ingress into concrete and therefore subsequent service life predictions [16].

When a crack of known geometry is introduced into such a model, the user must supply the diffusion coefficient to be used in the crack, as a function of crack width [4,11]. Alternatively, this diffusion rate could be assumed equal to that of chloride ions in free water, $1.8 \times 10^{-9} \text{ m}^2/\text{s}$ at 20°C [17]. In the current effort, ANSYS will be used to implement a simplified model for the reduction in service life due to the presence of a crack above the reinforcement, neglecting any contributions of binding or time dependency of the diffusion coefficient. This will be compared to a more complex and more realistic model implemented in the COMSOL commercial software package, as described below, to evaluate the importance of including binding rates (kinetics) and crack width into such simulations. In the ANSYS model, the crack (of depth T_{crack}) is simply modeled as an additional chloride containing boundary (with an unlimited supply of chloride ions) in a two-dimensional simulation and only half of the cracked system is modeled due to symmetry considerations (e.g., adiabatic boundary at crack central axis, Figure 2). This model can be executed for various combinations of reinforcement depth (T_{rebar}) and overall deck thickness (T_{concrete}) and compared to a version where the crack is simply omitted, so as to determine the reduction in service life produced by the crack's presence.

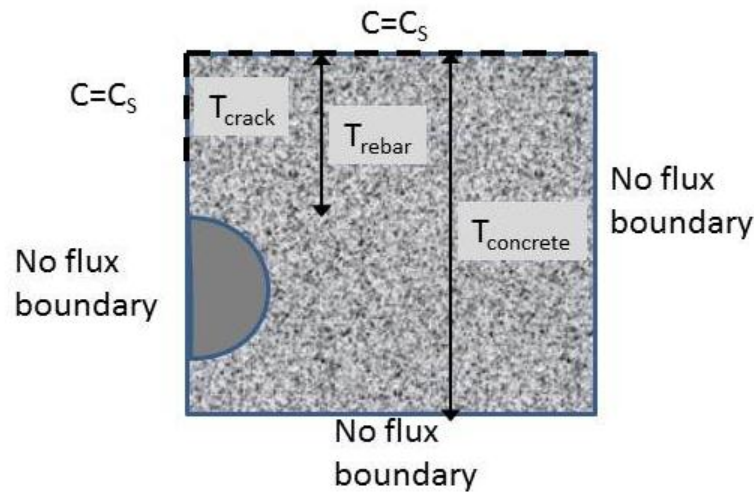


Figure 2. Simplified ANSYS geometry for simulating a single crack (2-D line boundary) in a two-dimensional concrete simulation.

It should be mentioned that in long term simulations, the boundary condition applied at the bottom surface of the concrete can have a significant influence on the buildup of chlorides at

the (top) depth of the steel reinforcement. If a no flux boundary condition is applied (as shown in Figure 2), the longtime equilibrium solution will be one in which the concentration is uniform throughout the depth of the concrete structure ($C=C_S$ at all depths). However, if a constant boundary condition of $C=0$ were to be instead applied at the bottom surface (implying, for example, that the bottom surface was washed periodically in an attempt to remove chlorides from that surface) the final equilibrium would be a linear profile of chlorides varying from $C=C_S$ at the top surface to $C=0$ at the bottom surface. In the relatively rare cases where the surface chloride concentration is only slightly higher than that required for reinforcement corrosion, such washing could potentially increase the service life of the concrete.

Typically accepted values for the chloride ion concentration necessary to initiate corrosion of steel reinforcement are on the order of 0.71 kg/m^3 to 1.2 kg/m^3 concrete for bare steel [18] and 2.1 kg/m^3 concrete to 4.3 kg/m^3 concrete for epoxy-coated reinforcement [19]. Surface concentrations are generally at least 8.3 kg/m^3 concrete [19], and can be two to three times higher [3,20]. Based on these accepted values, for most practical scenarios in the U.S., the ratio of the chloride concentration necessary to induce corrosion to that present in the environment (at the exposed concrete surface) is typically on the order of 0.1 to 0.3, such that reducing the chloride ion concentration at the bottom surface via washing will have a generally negligible influence on resultant service life, in terms of the top layer of steel reinforcement entering corrosion. For example, Figure 3 shows ANSYS simulation results obtained when applying a bottom boundary condition of either no flux or $C=0$ for 200 mm thick bridge decks with the top reinforcement present at a depth of either 100 mm or 50 mm and a 25 mm deep crack. For the reinforcement at 50 mm, the difference in chloride concentration only becomes significant for a C_{rebar}/C_S ratio of greater than 0.75. For the more extreme reinforcement depth of 100 mm, this value must be greater than approximately 0.4 to achieve a noticeable difference between the error function semi-infinite solution or the no flux boundary condition ANSYS simulation and the ANSYS simulation with a constant $C=0$ bottom surface boundary condition. Naturally, this bottom surface boundary condition would be influenced by the presence of stay-in-place metal forms, which have been observed to produce increased effective diffusion coefficients of concrete bridge decks in Utah [20], likely due to their higher moisture content.

COMSOL

The second simulation platform that was investigated is the COMSOL Multiphysics¹ package. This software contains specific modules for diffusion and absorption/reaction in multiple dimensions. This conveniently allows for the implementation of a sorption isotherm (linear in this case) for chloride interaction with the cementitious matrix, including the kinetics (first order rate process) of sorption. In the COMSOL simulations, an actual crack geometry (assumed rectangular) was simulated, as opposed to simply adding a one-dimensional chloride-containing boundary (line), as was done in the ANSYS simulations. In the COMSOL simulations, the user specifies the chloride diffusion coefficient in the bulk concrete, in a “damaged” zone of concrete near the crack [21], and within the crack itself. The parameters used in the COMSOL simulations, as obtained by calibration against an existing experimental data set (presented below) for cracked mortars [22], are summarized in Table 1. The purpose of the calibration is not to obtain a perfect fit to the experimental data, but rather to arrive at a set of ‘reasonable’ parameter choices that can then be employed in a parametric study to examine the

influence of crack geometry on relative service life for concrete bridge decks of known thickness and cover depth.

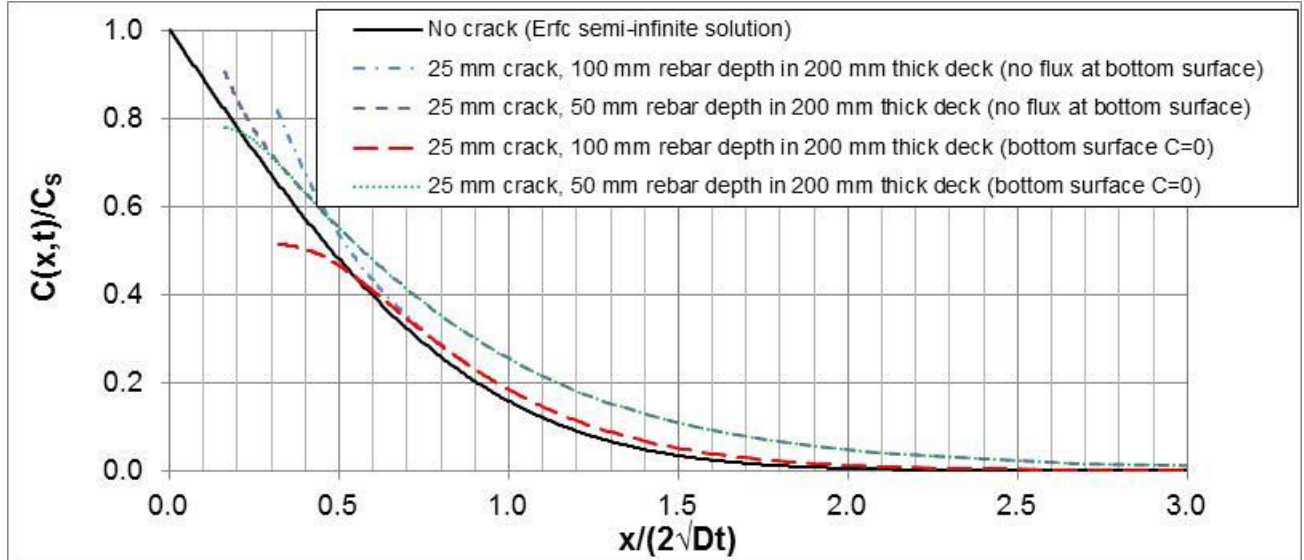


Figure 3. Normalized concentration vs. $x/(2\sqrt{Dt})$ for ANSYS simulated bridge decks with two different boundary conditions imposed at their bottom surface.

Table 1. Parameters used in COMSOL simulations

Parameter	Calibrated value
Diffusion coefficient of bulk mortar (D_{bulk})	$6 \times 10^{-11} \text{ m}^2/\text{s}$
Sorption rate constant of bulk mortar (k_{bulk})	$3 \times 10^{-7} \text{ s}^{-1}$
Diffusion coefficient of damaged mortar ($D_{damaged}$)	$1.2 \times 10^{-9} \text{ m}^2/\text{s}$
Sorption rate constant of damaged mortar ($k_{damaged}$)	$3 \times 10^{-7} \text{ s}^{-1}$
Diffusion coefficient in small crack, $CW^A < 100 \text{ } \mu\text{m}$	$2 \times 10^{-9} \text{ m}^2/\text{s}$
Diffusion coefficient in large crack, $CW \geq 100 \text{ } \mu\text{m}$	$4 \times 10^{-9} \text{ m}^2/\text{s}$
Alpha coefficient in linear isotherm (α)	4
External chloride concentration (3 % NaCl)	523.4 mol/m^3 pore solution
Porosity fraction of mortar	0.1828
Size of damaged zone for small crack, $CW < 100 \text{ } \mu\text{m}$	1 mm
Size of damaged zone for large crack, $CW \geq 100 \text{ } \mu\text{m}$	4 mm

^A CW is measured crack width in micrometers [22]

While Freundlich, Langmuir, and linear isotherms [23] have all been employed in past studies of chloride binding in concrete, for these simulations, a simple linear isotherm was utilized to express the relationship between free and bound chlorides:

$$C_{bound} = \alpha C_{free} \quad (2)$$

where α is a dimensionless fitting parameter, here taken as 4, as obtained by fitting the experimental data (30 d and 90 d exposure) for a mortar specimen with no cracks [22]. This is the best fit value for equation 2 when C_{free} and C_{bound} are both expressed in units of $\text{mol Cl}^-/\text{m}^3$ pore solution. Since the experimental results in [22] are presented in terms of total chlorides

measured as a mass percentage of mortar, the estimated C_{free} and C_{bound} values are multiplied by the molecular mass of chloride (0.03545 kg/mol) and the estimated total porosity of the mortar (18.3 %) and then divided by the density of the mortar (2301.4 kg/m³ based on the reported mortar mixture proportions [22]). In this study, this single isotherm equation was utilized to model the combined effects of sorption of the Cl⁻ ions by the C-S-H and other hydration products, as well as the reaction of these ions with aluminate phases to form Friedel's salt (and perhaps other chloroaluminate compounds).

This sorption/reaction is implemented in COMSOL by adding a first-order reaction term to the standard diffusion equation, resulting in:

$$\frac{\partial C}{\partial t} = \nabla \cdot (D \nabla C) + k * (C_{bound} - \alpha C) \quad (3)$$

Here, the driving force for binding is the difference between the equilibrium bound chloride and its current value, e.g. ($C_{bound} - \alpha C$). Using equation 3 and the sorption rate constant, k_{bulk} , provided in Table 1, the simulation results for the uncracked mortar after 30 d and 90 d of chloride exposure are compared to the experimental values [22] in Figure 4; in this case, a reasonable, but not perfect, agreement is obtained between experiment and simulation. Without including the sorption/reaction effects, the experimental curves shown in Figure 4 were previously fitted to equation 1 [22], resulting in diffusion coefficients of 2.34×10^{-11} m²/s and 1.09×10^{-11} m²/s for the 30 d and 90 d results, respectively. While the diffusion coefficient could be decreasing with time due to continued hydration, the 3 % NaCl solution ponded on top of the mortar specimens translates into a mortar chloride content of only about 0.15 %, significantly lower than the projected y-axis intercepts for the two experimental curves in Figure 4. This difference in intercept can be better explained by the binding/reaction of the chlorides within the cementitious matrix of the mortar, as the results shown in Figure 4 are for total (and not free) chloride concentrations. For the COMSOL simulations, once an 'appropriate' selection was made for the binding isotherm and its rate constant (Table 1), a reasonable fit to the two experimental data sets was obtained by using a single value for the diffusion coefficient that does not decrease over time, effectively assuming that the 43 d of curing applied in [22] was sufficient to hydrate a majority of the available cement. The importance of binding/reaction in regulating chloride ion penetration is indicated in Figure 5 by the bound chloride concentration being significantly higher than the free chloride concentration after only 30 d of chloride exposure.

NIST model

A third model was developed at the National Institute of Standards and Technology (NIST) based on a generalized finite difference solution of the diffusion/binding problem in three dimensions [24] and also incorporating the kinetics of the chloride ion binding/reaction. Unlike the COMSOL model, the NIST model features a single element size and does not include any variable mesh refinement algorithm(s) to reduce memory requirements and computational times. Results of this model were compared with the COMSOL-generated results (Figures 4 and 5) to provide further confidence that the simulations using the two packages were being conducted correctly. As shown in Figures 4 and 5, although different meshes and solvers are employed in each case, the NIST and COMSOL models produce basically identical results for the case of the uncracked mortar, when provided with the same diffusion coefficient and binding/reaction

parameters. Both models also successfully reproduce Fickian behavior (in comparison to equation (1)) when executed with their binding/reaction terms removed.

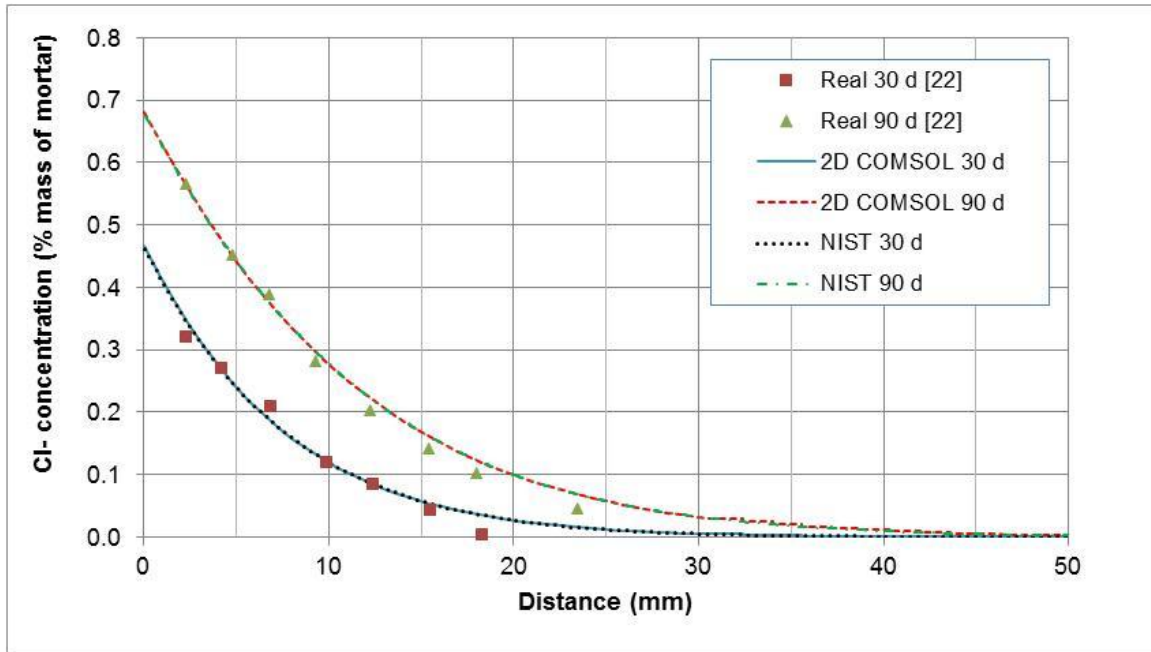


Figure 4. Comparison of COMSOL and NIST-simulated total chloride profiles with experimental data obtained directly from graphs presented in [22]. The experimental data points represent the chloride contents determined from fixed volume physical samples obtained at various depths. The lines for the COMSOL and NIST models basically overlap one another.

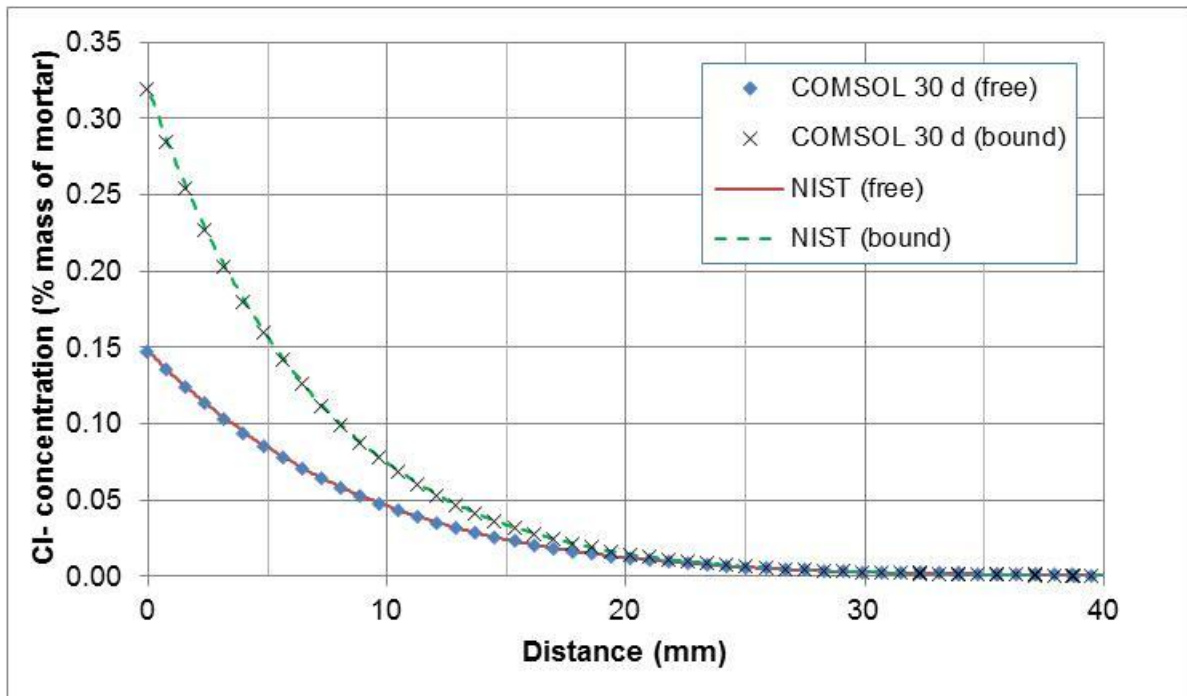


Figure 5. COMSOL and NIST-simulated free and bound chloride profiles after 30 d of chloride exposure.

Results

Calibration of COMSOL model for cracked mortars

As with any modeling effort, assumptions were made in developing the COMSOL simulations for comparison with experimental data [22]. These fitting hypotheses were obtained over the course of a trial-and-error fitting process and are detailed as follows:

- 1) In addition to the bulk mortar and the crack itself, a third zone consisting of ‘damaged’ mortar surrounding the crack is modeled. For crack widths less than 100 μm , the damaged zone has a dimension of 1 mm, while for crack widths greater than or equal to 100 μm , it is set at 4 mm. Win et al. [21] had previously suggested the existence of such a damaged zone, several mm in extent, around the crack, based on their observation of a greater penetration depth of chlorides spreading from the crack than that observed spreading from the top (uncracked) portion of their exposed specimens. In COMSOL, the damaged zone has the same porosity and sorption characteristics as the bulk mortar, but a diffusion coefficient, D_{damaged} , that was adjusted to be twenty times higher than the corresponding rate in the bulk mortar (Table 1); this increase could potentially be due to microcracking and an accompanying increase in capillary pore and interfacial transition zone connectivity.
- 2) For crack widths greater than 100 μm , the diffusion coefficient of chloride ions within the crack itself is doubled from $2 \times 10^{-9} \text{ m}^2/\text{s}$ to $4 \times 10^{-9} \text{ m}^2/\text{s}$ to empirically allow for the expected increased transport of chloride ions due to the (gravity induced) convection of a higher density (i.e. higher Cl^- concentration) fluid from above the crack into the crack, where the local concentration of Cl^- solution has been decreased due to diffusion of Cl^- ions into the surrounding mortar. In nature, this would be similar to the sinking of higher concentration salt water towards the ocean floor, resulting in the formation of ice ‘brinicles’ in arctic waters [25].
- 3) The modeled total chloride concentrations within the cylindrical (15 mm) core evaluated in the experiments were fitted to a quadratic function (Appendix A) for the purpose of comparing to the experimentally measured chloride contents on a mass of mortar basis for cored cylinders from the cracked mortar beams (it is further assumed that the cylindrical cores were taken perfectly centered on the crack). Additionally, these model values were further increased to include the free chlorides that would be present in the solution in the crack, assuming that these chlorides would be uniformly deposited on the side crack walls, as the solution evaporates at the end of the chloride exposure (prior to the coring).

Employing these assumptions, simulations were executed for the six different cracked specimens prepared and evaluated in reference [22]. Figure 6 provides example two-dimensional concentration maps for three systems with different crack widths, while Figure 7 provides a comparison of the experimental data presented in [22] to the COMSOL simulation results for the 76 mm thick specimens, with a no flux boundary condition applied at their bottom surface. While far from perfect, the agreement observed in Figure 7 is reasonable given the assumptions in the simulations and the unknowns and variability in the experiment. A better agreement can be obtained by making the damaged zone size different for each crack or even removing it entirely in the case of the smallest crack, but it was felt that having a damaged zone that is the same for ‘small’ and ‘large’ cracks made more sense with respect to the ultimate purposes of this study, namely to discern the influence of cracking on service life. Still, it is clear from both the experimental and simulation results in Figures 6 and 7 that crack geometry has a major influence

on chloride penetration, with the expected observation that wider, deeper cracks are significantly more detrimental. Previously, several research groups have suggested that cracks below about 80 μm in width have minimal influence on chloride ingress [4,12,13]. To bracket this performance in the simulations of service life reduction to follow, extremes of a 50 μm wide, 25 mm deep, and a 500 μm wide, 45 mm deep crack will be investigated. The crack depths selected for these two crack widths are in general agreement with the experimental data presented in [22] and also with theoretical analysis [26] and experimental data of Weiss et al. [27].

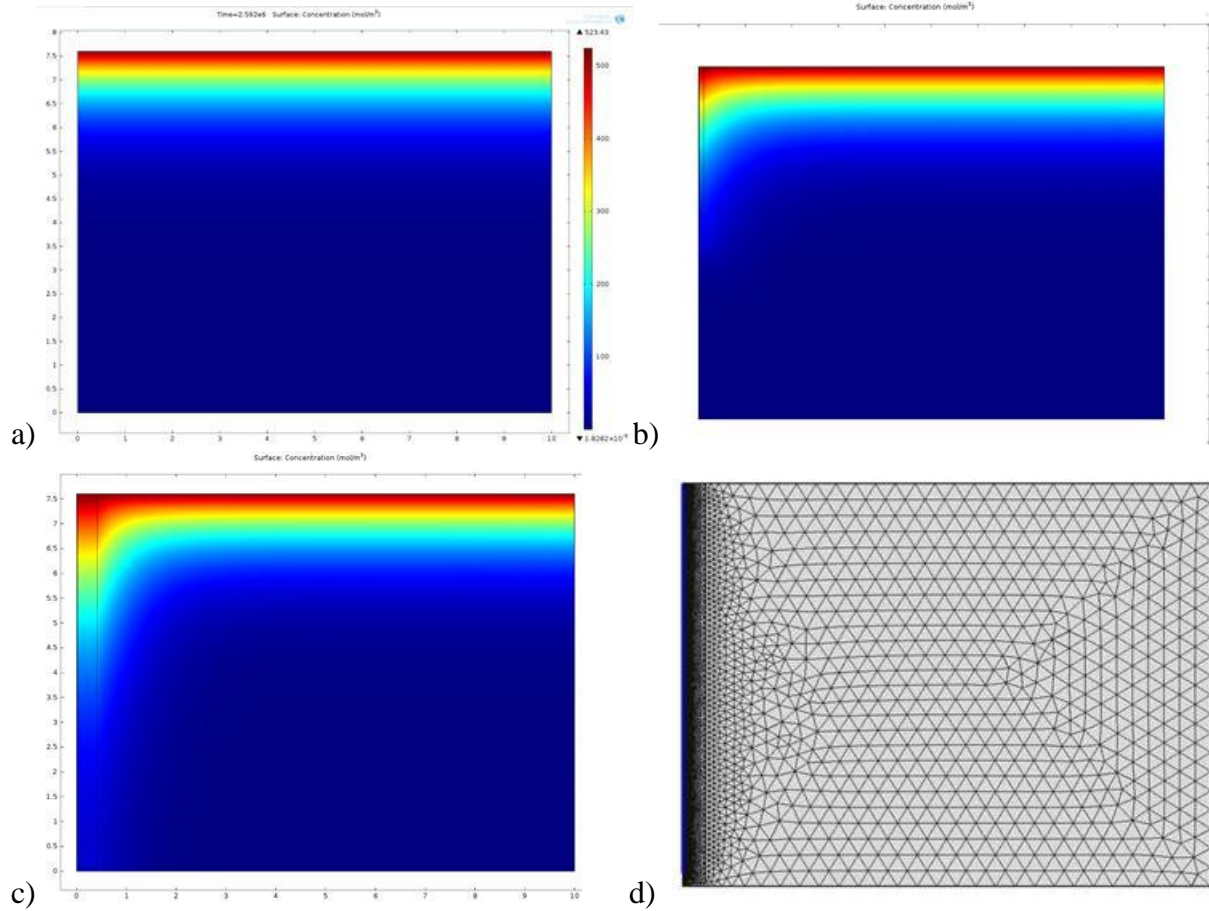


Figure 6. Color-coded concentration maps (red is high and blue is low concentration) for three COMSOL two-dimensional simulations of a concrete slab at 30 d: a) No crack, b) 102.9 μm wide by 36.6 mm deep crack with a 1 mm wide damaged zone, and c) 392 μm wide by 73.4 mm deep crack with a 4 mm wide damaged zone, along with d) a view of the COMSOL variable meshing for the 392 μm wide by 73.4 mm deep crack. The vertical axis indicates the depth into the concrete while the horizontal axis indicates the distance from the (cracked) center of the slab.

In each case, when present, the crack is located at the upper left hand corner of the image.

Influence of cracking on projected concrete service life

Using the parameters determined in the above calibration process, simulations were conducted using both ANSYS and COMSOL to estimate the decrease in concrete service life due to cracking as a function of crack geometry and concrete cover depth for a series of 200 mm thick bridge decks, sealed at their bottom surface (i.e. no flux). Figure 8 provides the results of

the ANSYS simulations for the no binding case, while those from COMSOL that include binding/reaction are provided in Figures 9 and 10 (two different values for diffusion coefficients).

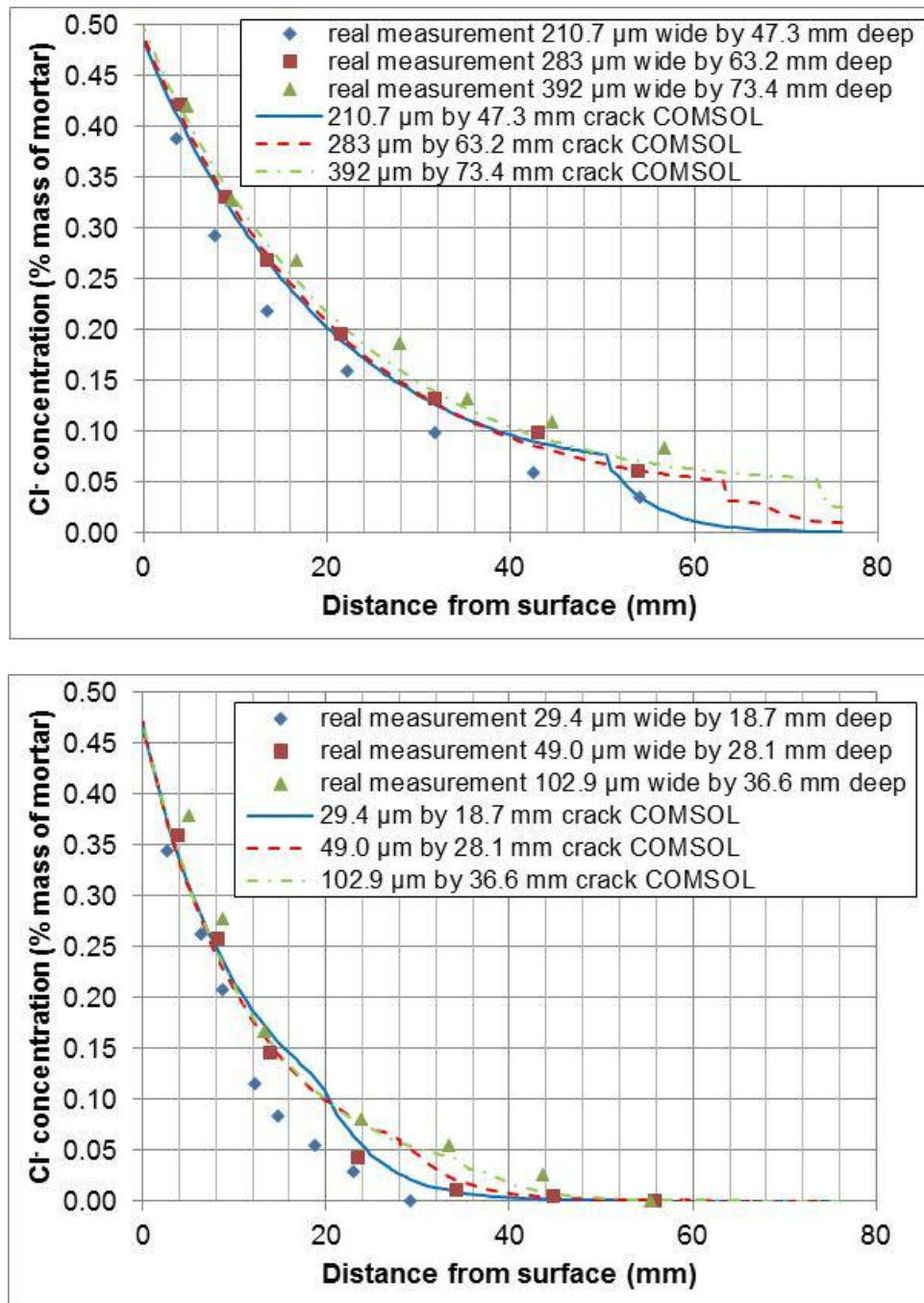


Figure 7. Comparison of COMSOL simulations to experimental total chloride profiles [22] for large (top) and small (bottom) cracks.

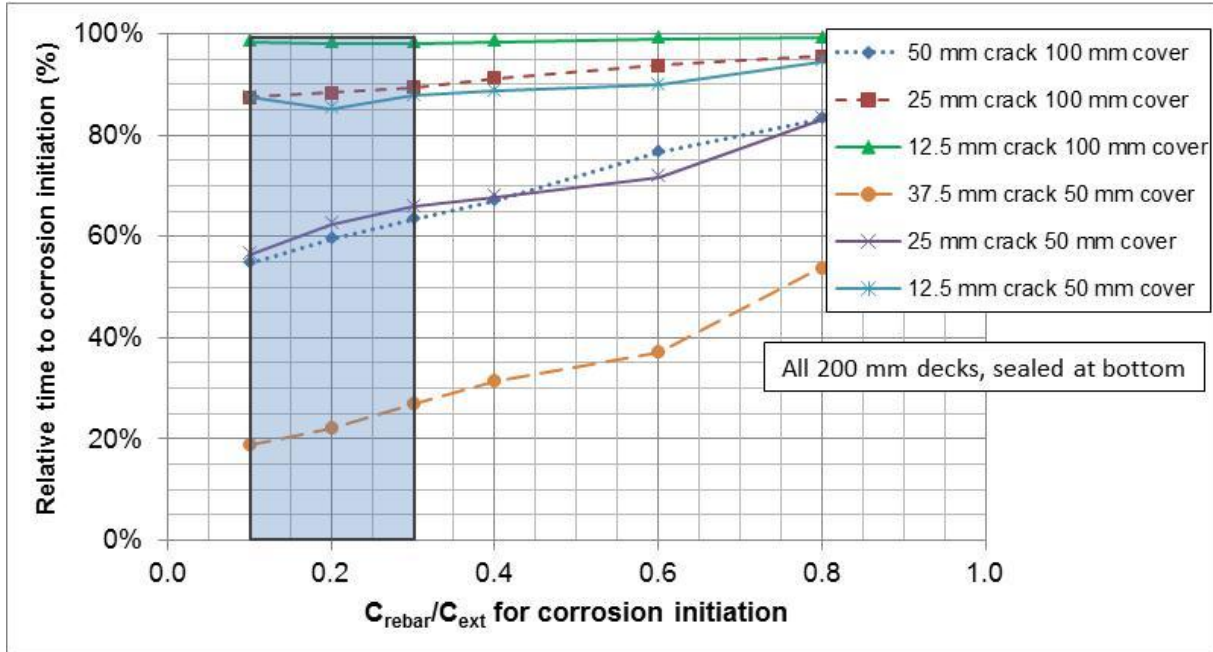


Figure 8. Projected reductions in service life for cracked concrete bridge decks as simulated using ANSYS (diffusion only). The highlighted area indicates the results for the ratios of C_{rebar}/C_{ext} of 0.1 to 0.3, commonly expected for concrete in the field.

To generate the curves shown in Figure 8, the ANSYS simulations were executed for a time period of 40 years (with $D=2 \times 10^{-11} \text{ m}^2/\text{s}$). The results in Figure 8 are controlled by the form of equation (1) for pure diffusion with no binding in a semi-infinite media, as modified slightly by the no flux boundary condition applied at the bottom surface. As such, the results nearly overlap for fixed ratios of the crack depth to the cover depth (e.g., 1:4 and 1:2 shown in Figure 8). It is well known that for one-dimensional Fickian diffusion, halving the cover depth will reduce the projected service life by a factor of four [15]. Because the crack is effectively a line source oriented perpendicular to the top rebar plane in these two-dimensional simulations, the reduction in service life is significantly reduced relative to that produced by an equivalent decrease in the cover depth (with the new cover depth computed as the reinforcement depth – the crack depth for example). For the range of C_{rebar}/C_{ext} of 0.1 to 0.3 that is most relevant to typical field exposures, to first order, the relative reduction in service life is instead slightly less than or equal to the relative reduction in cover computed based on the distance from the crack tip to the top of the reinforcement. For example, a crack that penetrates halfway to the top level of the steel reinforcement results in a service life reduction of slightly less than 50 % (projected relative service life of 55 % to 65 % in Figure 8).

Figure 8 could be used with existing models to estimate the impact of measured or anticipated cracking on projected concrete service life. For example, if a current service life model projects a service life of 100 years for a 200 mm thick bridge deck with 50 mm concrete cover and the deck is found to contain traverse cracks that penetrate to a depth of 25 mm, the estimated service life would be adjusted to a range of 57 years to 66 years for C_{rebar}/C_{ext} ranging from 0.1 to 0.3, according to the points shown on the graph in Figure 8. If epoxy-coated rebar were used and the value of C_{rebar}/C_{ext} was increased to 0.6, for example, the expected service life would only be reduced to 72 years. Figure 8 (as well as Figures 9 and 10) indicates that field

exposures with higher values of C_{rebar}/C_{ext} are less sensitive to the presence of cracks. Of course, there are two ways to increase the value of this ratio, either by increasing the threshold value of C_{rebar} necessary to initiate corrosion via epoxy-coated rebar, corrosion inhibitors, etc. or by decreasing the value of C_{ext} , typically by decreasing the amount of chlorides applied to the concrete surface during the winter months or switching to a non-chloride-based source of deicing chemicals.

The ANSYS analysis presented in Figure 8 did not include the presence of binding/reaction or any influence of the crack width, as the cracks were simply modeled as an additional surface providing a constant chloride concentration. This simplistic analysis can be contrasted against the more rigorous evaluation using the COMSOL model that is presented in Figures 9 and 10. For the results presented in Figure 9, the diffusion coefficient of the bulk concrete was set at $6 \times 10^{-11} \text{ m}^2/\text{s}$, the same value as that in Table 1, obtained for the mortar specimens [22] analyzed for model calibration in this study. For those in Figure 10, this value was reduced by a factor of ten to $6 \times 10^{-12} \text{ m}^2/\text{s}$ to simulate the influence of cracks on a high performance concrete that typically has a much lower (uncracked) diffusion coefficient. In the latter case, the value of the diffusion coefficient in the damaged zone in the cracked systems was also reduced from the value provided in Table 1 by a factor of ten to $1.2 \times 10^{-10} \text{ m}^2/\text{s}$. Due to these much lower diffusion coefficients, the simulations for Figure 10 had to be executed for a simulated time period of 400 years.

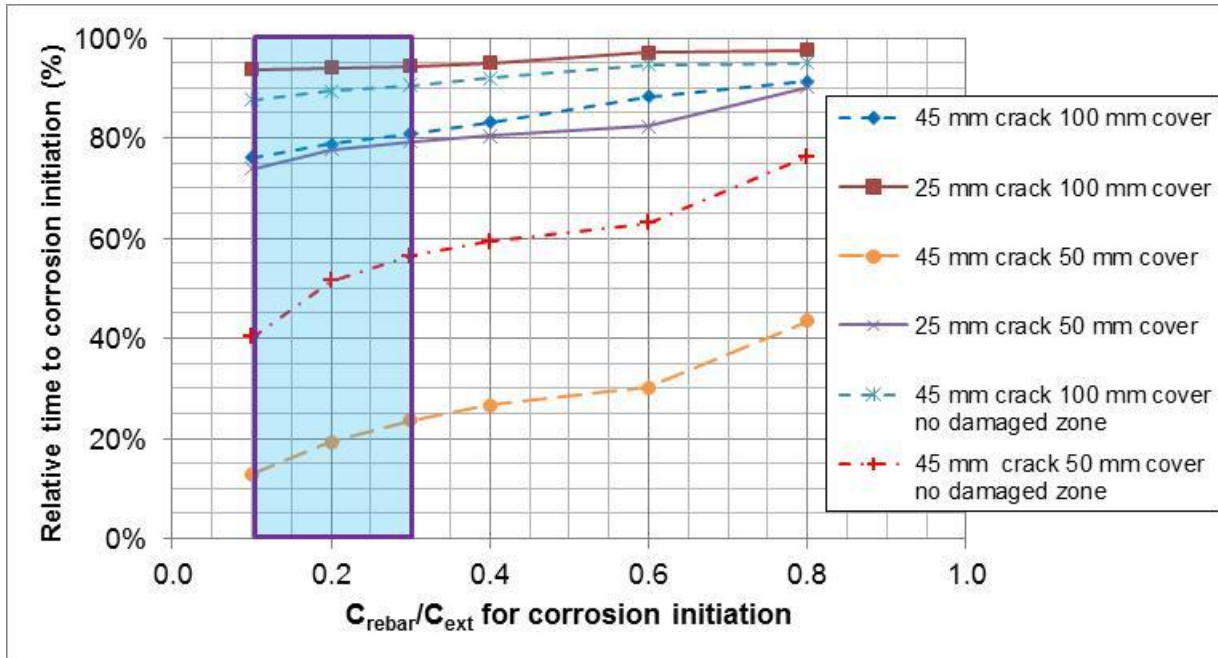


Figure 9. Projected reductions in service life for cracked concrete bridge decks as simulated using COMSOL (diffusion and binding) with $D_{conc}=6 \times 10^{-11} \text{ m}^2/\text{s}$.

The results in Figure 9 indicate that binding and its kinetics have a significant impact on the influence of cracking on projected service life. In comparing the 25 mm crack systems in Figure 9 (with $D_{conc}=6 \times 10^{-11} \text{ m}^2/\text{s}$) to those in Figure 8, for example, it is clearly observed that the reduction in projected service life is significantly less when the binding of chlorides is included in the simulations and when the crack is modeled by its actual geometry as opposed to

being an additional line source of chlorides. As the binding removes chlorides from the pore solution, it helps mitigate the negative impact of the cracking on chloride penetration. For the example of a current service life model projecting a service life of 100 years for a 200 mm thick bridge deck with 50 mm concrete cover, when the deck is found to contain traverse cracks that penetrate to a depth of 25 mm, the estimated service life would be adjusted to a range of 74 years to 79 years for C_{rebar}/C_{ext} ranging from 0.1 to 0.3, versus the range of 57 years to 66 years determined above for the ANSYS diffusion only simulations. In this case, employing the simplified diffusion only analysis with an additional line source of chlorides would underestimate the projected service lives by about 15 years or about 25 %.

In Figure 9, additional results are presented for the case of the larger crack, but without any accompanying damaged zone. The impact of the anticipated damaged zone (assigned to be 4 mm in extent for the larger crack in the current simulations) is significant, as the projected service life is reduced dramatically when the damaged zone is included in the simulation. As would be expected, for the case of a 45 mm deep crack with only a 50 mm cover depth, the reduction in service life produced by the addition of a 4 mm damaged zone is particularly large, with relative times to corrosion being reduced from a range of 40 % to 57 % to one of 12 % to 24 % for C_{rebar}/C_{ext} ranging from 0.1 to 0.3.

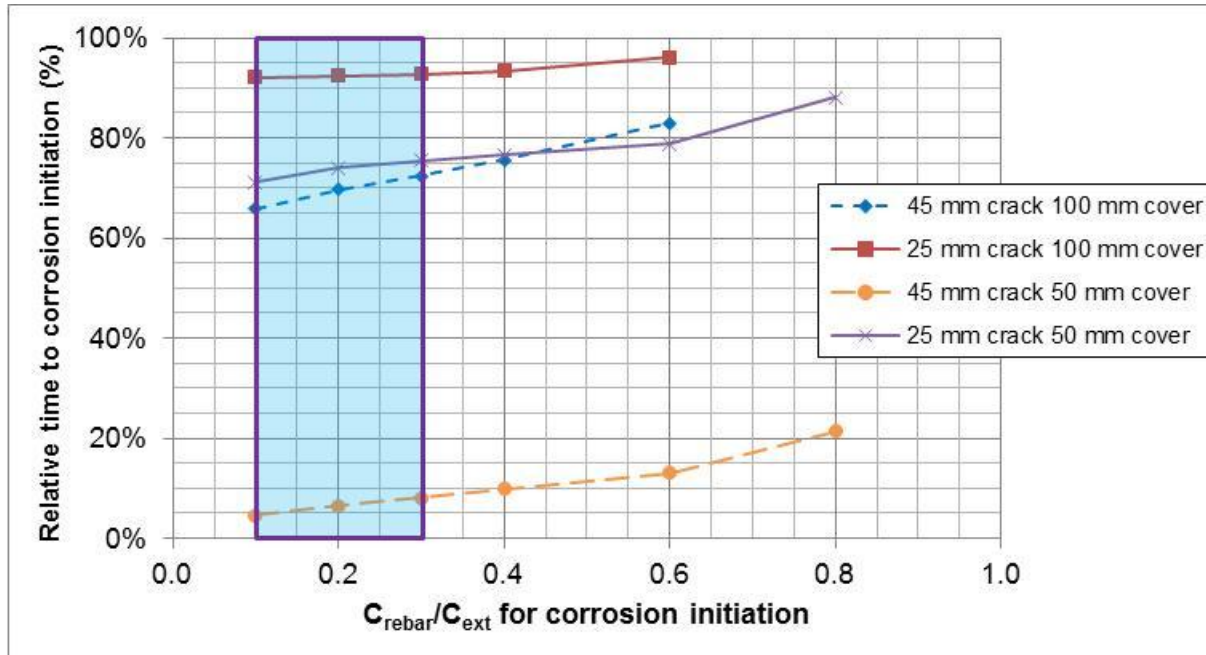


Figure 10. Projected reductions in service life for cracked concrete bridge decks as simulated using COMSOL (diffusion and binding) with $D_{conc}=6 \times 10^{-12} \text{ m}^2/\text{s}$.

In contrasting Figures 9 and 10, it is observed, as might be expected, that the relative influence of cracks become more dramatic in the case of high performance concrete with a lower diffusion coefficient. For the 200 mm thick bridge deck with 50 mm cover depth, for example, the estimated service life range is 72 years to 75 years, moving from the range of 74 years to 79 years for the higher D_{conc} COMSOL simulation towards the ANSYS/diffusion only range of 57 years to 66 years. Even when considering a high performance concrete, the diffusion only

(ANSYS) analysis is conservative in over predicting the projected service life reduction due to the presence of cracking.

As illustrated by the 45 mm depth crack with 50 mm cover, as cracks approach the reinforcement depth, their relative influence in a high performance concrete becomes even more significant. In this case, the high performance concrete service life (Figure 9) is reduced to only 5 % to 8 % of its initial (uncracked) value, while the ordinary concrete service life (Figure 8) is only reduced to about 12 % to 24 % of its initial (but lower) value. Thus, avoiding or minimizing cracks is seen to be particularly important in applications employing high performance concrete to achieve a long service life.

Conclusions

A comprehensive approach for estimating the influence of transverse cracking directly above the steel reinforcement on chloride penetration and concrete service life has been presented, concluding that:

- A simple 2-D simulation that considers only diffusion can be used to provide a conservative estimate on the influence of cracking on projected concrete service life.
- Because of the significant impact of chloride binding on chloride penetration into concrete, more accurate estimates of the influence of cracking on chloride penetration and service life require a simulation that includes the influence of binding and its kinetics. Such a simulation can be developed from the ground up based on a set of diffusion and reaction equations, or adapted from existing commercial simulation software packages (such as COMSOL).
- The incorporation of binding into such models increases the projected service life of the cracked concrete, relative to that predicted by a model that only includes diffusion.
- Cracking has a relatively larger impact on chloride penetration in high performance concrete systems. Unfortunately, high performance concretes are often more prone to cracking than conventional concretes. It must be noted, however, that the present study did not consider whether cracks in high performance concrete may have different dimensions (e.g., being narrower or even many times wider) and damaged zone characteristics than their counterparts in ordinary concrete.

This analysis considered only cracks that penetrate partway to the depth of the reinforcement, nominally considering any cracks that do reach the reinforcement depth to be sufficient to allow immediate chloride access to the bar. It is recognized that additional efforts in characterizing what comprises ‘typical’ cracking for field concretes (number, size, and depth) are sorely needed. Additionally, it is recognized that the simulations in this paper were developed for concretes under continuously saturated conditions. That is one of the reasons that the graphs in Figures 8 to 10 were developed to express the *relative* reduction in concrete service life, relative to that predicted by any existing concrete service life model. These existing models may or may not include the influence of partial saturation (wetting/drying) on concrete service life. It is proposed that the presented graphs may be used with any of these existing models to estimate the projected relative reduction in service life due to the presence of cracking, whether these cracks are measured ones from existing structures or hypothetical ones. In the future, more complex

simulations of chloride penetration that include both cracking and wetting and drying exposures, among other factors, could be developed using COMSOL or other simulation platforms.

Appendix A – Derivation of Cylindrical Sampling Average Concentration

In many real world studies, a circular drill bit is used to obtain a series of powder samples at different depths from a concrete surface, which are then analyzed for their chloride contents. For a one-dimensional diffusion scenario in a homogeneous media, the measured concentration, per gram of powder for example, accurately represents the one-dimensional diffusion profile. However, in the case of a cracked specimen that is cored “through” the crack, the two-dimensional boundary conditions established for the diffusion create variations in chloride concentrations in two dimensions, both as a function of depth and as a function of distance from the crack. To compare numerical and experimental results in this case, the numerical profiles must be properly integrated over the circular sampling area. A procedure for estimating these values from a set of three simulated linear profiles measured at different distances from the crack center is presented here.

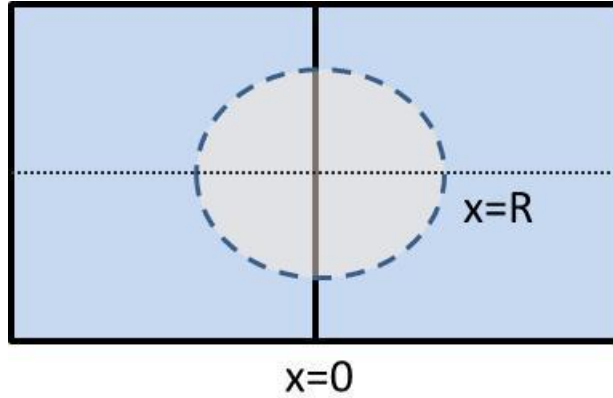


Figure A.1 Schematic of cored region using a circular drill bit, centered on the crack.

Consider a circular sampling bit of radius R , centered on a straight crack in the y -direction as shown in Figure A.1. Due to symmetry considerations, one can consider only a quarter of the circular sample, with an area of $\pi R^2/4$. Consider three profiles measured as a function of depth at different distances from the crack center in the x -direction, $f(x=0, y=0)$, $f(x=R/2, y=0)$, and $f(x=R, y=0)$. If one assumes a quadratic fit for the concentration as a function of this distance in the x -direction ($f(x)=a+bx+cx^2$) for this concentration data, then

$$f(0, 0) = a \quad (\text{A.1})$$

$$f\left(\frac{R}{2}, 0\right) = a + b\left(\frac{R}{2}\right) + c\left(\frac{R^2}{4}\right) \quad (\text{A.2})$$

$$f(R, 0) = a + bR + cR^2 \quad (\text{A.3})$$

Considering the function for the x coordinate only and solving for a , b , and c yields

$$a = f(0) \quad (\text{A.4})$$

$$b = \frac{4}{R} \left[f\left(\frac{R}{2}\right) - \frac{3}{4}f(0) - \frac{1}{4}f(R) \right] \quad (\text{A.5})$$

$$c = \frac{2}{R^2} \left[f(R) + f(0) - 2f\left(\frac{R}{2}\right) \right] \quad (\text{A.6})$$

Next, one needs to integrate this concentration function $(a+bx+cx^2)$ over the quarter circle and divide by the area of the quarter circle to get the average (measured) concentration:

$$\bar{C} = \frac{\int_0^R (a+bx+cx^2)\sqrt{R^2-x^2}dx}{\frac{\pi R^2}{4}} \quad (\text{A.7})$$

From the handbook of mathematics [28],

$$\frac{\int_0^R a\sqrt{R^2-x^2}dx}{\frac{\pi R^2}{4}} = a \quad (\text{A.8})$$

$$\frac{\int_0^R bx\sqrt{R^2-x^2}dx}{\frac{\pi R^2}{4}} = \frac{4R}{3\pi}b \quad (\text{A.9})$$

and

$$\frac{\int_0^R cx^2\sqrt{R^2-x^2}dx}{\frac{\pi R^2}{4}} = \frac{R^2}{4}c \quad (\text{A.10})$$

Finally, by substituting these integrals, along with the values for a , b , and c into the average concentration expression in equation A.7, the average concentration is given by

$$\bar{C} = f(0) + \frac{16}{3\pi} \left[f\left(\frac{R}{2}\right) - \frac{3}{4}f(0) - \frac{1}{4}f(R) \right] + \frac{1}{2} \left[f(R) + f(0) - 2f\left(\frac{R}{2}\right) \right] \quad (\text{A.11})$$

Note that this value correctly reduces to $f(0)$ when there is no crack (i.e., when $f(0)=f(R/2)=f(R)$).

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